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## New Materials from High Pressure

High pressure methods are important for synthesising new materials, and exploring changes of structure and property in dense matter. High pressure materials science will be introduced briefly, and applications for materials chemistry will be illustrated with reference to new oxides and nitrides. High pressure often stabilises cations in unusual oxidation or coordination environments. Examples are perovskites with  $\text{Mn}^{2+}$  at A-sites, such as  $\text{MnVO}_3$  [1], the double perovskite  $\text{Mn}_2\text{FeReO}_6$  [2] and double double perovskites  $\text{MnRMnSbO}_6$  and  $\text{CaMnFeReO}_6$  with order of A and B site cations [3,4,5]. A remarkable variety of new iron oxides has recently been reported at high pressures, and we have explored the substitutional chemistry of  $\text{Fe}_4\text{O}_5$ . Complex magnetic orders are observed in  $\text{MnFe}_3\text{O}_5$  [6] and  $\text{CoFe}_3\text{O}_5$  [7], while  $\text{CaFe}_3\text{O}_5$  (prepared at ambient pressure) shows electronic phase separation driven by trimeron formation [8]. A new quantum phenomenon, quantised weak ferromagnetism, has recently been discovered in the perovskite  $\text{YRuO}_3$  based on the unusual  $\text{Ru}^{3+}$  state [9]. A high pressure method using sodium azide has recently been developed to synthesise nitrides in high oxidation states giving the iron(IV) nitride,  $\text{Ca}_4\text{FeN}_4$  [10], an electron-localised  $\text{Ni}^{2+}$  nitride,  $\text{Ca}_2\text{NiN}_2$  [11], and a rare example of a nitride perovskite,  $\text{LaReN}_3$  [12]. The latter material can be decomposed to give novel reduction products  $\text{LaReN}_{2.5}$  and layered  $\text{LaReN}_2$  demonstrating topotactic reduction chemistry analogous to that of perovskite oxides like  $\text{LaNiO}_3$  and  $\text{SrFeO}_3$ .

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**Paul Attfield** holds a Chair in Materials Science at Extreme Conditions at the School of Chemistry and Centre for Science at Extreme Conditions, University of Edinburgh. He received B.A. and D.Phil. degrees from Oxford University, and he was a Co-Director of the Interdisciplinary Research Centre in Superconductivity at the University of Cambridge during 1991-2003. He received the Royal Society of Chemistry's Meldola and Corday-Morgan medals and Peter Day award, and he was elected a Fellow of the Royal Society in 2014. Early research contributions included pioneering resonant X-ray scattering experiments of cation and valence ordering, and studies of disorder effects in functional oxides. Current research is centred on electronic and magnetic materials including use of high pressure methods.